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An Effective Strategy for the Preparation of α,β -Unsaturated Hydrazones and Pyrazole Derivatives. Synthetic Applications of β -Functionalized Phosphorus Compounds.

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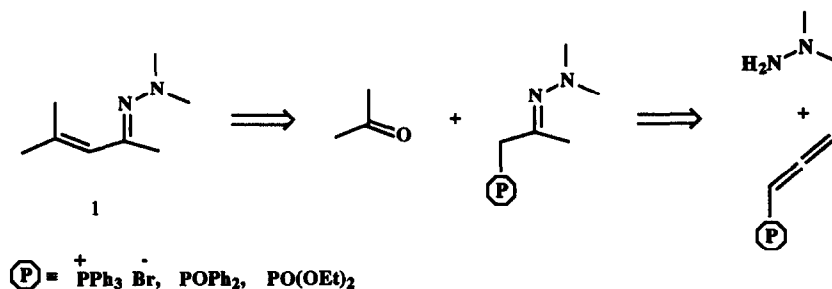
Abstract: β -Enehydrazino phosphonium salts **3** as well as β -hydrazono phosphine oxides **6** and phosphonates **7** are obtained from hydrazines, propargylphosphonium salts **2** and phosphorylated allenes **4** and **5**. β -Functionalized compounds **3**, **6** and **7** are used for the synthesis of α,β -unsaturated hydrazones **1**, pyrazoline **14** and pyrazole derivative **15**.

Hydrazones are common nitrogen derivatives of ketones and aldehydes and have received much attention in recent years because of their range of applications¹. They form part of the structure of new azapeptides² and of biologically active antibiotic compounds such as cirratiomycin^{3a}, antrimycin^{3b}, azinothrycin^{3c}, citropeptin^{3d} and megamycin^{3e}. Hydrazones can be also used as a protective group of the carbonyl function⁴, as formyl^{5a} and acyl^{5b} anion equivalents and synthetic intermediates in the preparation of heterocycles⁶, nitriles⁷, gem-difluorocompounds⁸ (which play an important role in biological chemistry⁹) as well as in the asymmetric synthesis of chiral amines^{10a}, α -aminoaldehydes^{10b} and α -aminoacids^{10c}. Likewise, a very important methodology for the formation of carbon-carbon bonds has been developed in recent years using carbanions derived from hydrazones¹¹, which leads to very successful applications in the enantioselective synthesis of oxosulfones^{12a}, pheromones^{12b}, the potassium channel opener RP66471^{12c} and of natural products such as the ionophore antibiotic indanomycin^{13a}, the sex pheromone serricornin^{13b} and the sesquiterpene (+) eremophilinolide^{13c}.

In the chemistry of hydrazones the usefulness of the of α,β -unsaturated hydrazones is particularly significant as a result of their potential as starting materials in the preparation of β -hydroxy-¹⁴ and α,β -unsaturated ketones¹⁵ as well as of biologically active pyrazoles¹⁶. Likewise, unsaturated hydrazones have

recently been shown to be a versatile tool for the construction of six-membered heterocycles by means of the Diels-Alder reactivity of these substances as 1-azadienes¹⁷. The lack of general methods for synthesis of these compounds has probably limited their use in organic synthesis. Simple α,β -unsaturated hydrazones are mostly synthesized by the condensation reaction of carbonyl compounds with hydrazines. However, the preparation of such compounds is far from simple and especially in the case of ketones, only yields good results in very specific cases and generally leads to Michael addition¹⁸.

In connection with our interest in the use of new β -functionalized phosphorus derivatives as synthetic intermediates in the preparation of acyclic¹⁹ and cyclic²⁰ derivatives, we have recently used phosphorus compounds as homologation reagents²¹ for conversion of carbonyl derivatives into α,β -unsaturated hydrazones with the introduction of two additional carbon atoms in the resulting chain. Here we aim to extend this methodology to the preparation of a wide range of unsaturated hydrazones **1** and to explore the synthetic use of phosphorylated hydrazones in the preparation of new groups of acyclic and cyclic compounds. Retrosynthetically, we envisaged obtaining hydrazones **1** by an olefination reaction of β -hydrazono phosphorus compounds (or their synthetic equivalents the tautomeric enehydrazino derivatives), obtaining both these from the addition of hydrazines to phosphorylated allenes (or the synthetic equivalent the propargylic phosphonium salts) (see scheme 1).



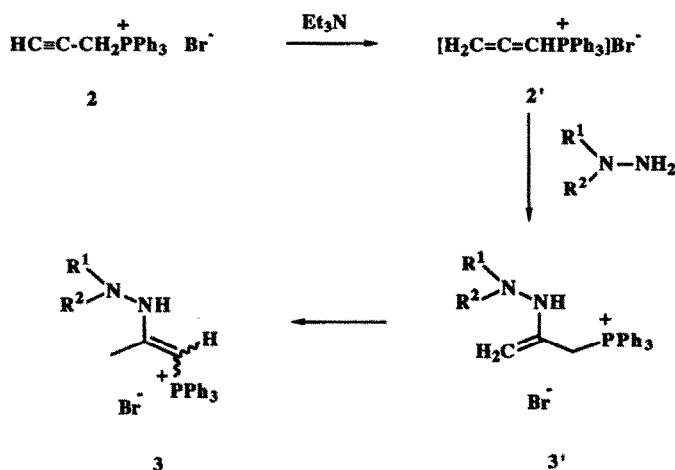
Scheme 1

RESULTS AND DISCUSSION

Preparation of β -enehydrazino phosphonium salts **3**.

The required β -enehydrazino phosphonium salts **3** were very easily prepared in high yields through the addition of achiral and chiral hydrazines to commercially available propargyltriphenylphosphonium bromide **2** in refluxing chloroform (Scheme 2, Table 1). Compounds **3** were characterized on the basis of their spectroscopic data, which indicate that they are isolated as a mixture of the *Z*- and *E*-substituted phosphonium salts **3**, although for our purposes the separation of *Z*- and *E*-isomers is not necessary for subsequent reactions. Thus, the ³¹P-NMR spectrum for **3a** showed two different absorptions at δ_p 11.8 and 17.4 ppm in an approximate isomer ratio 25 / 75 as evidenced by the relative peak areas for each salt, in which the high-field chemical shift corresponds to the *Z*-isomer **3a**. Further examination of the ¹H and ¹³C-NMR spectra is consistent with the enehydrazine structure of the phosphonium salts. In the ¹H-NMR spectrum of **3a**, the vinylic proton resonates at δ_H 4.61 as a well resolved doublet with coupling constant of ²J_{PH} 18 Hz. and the methyl

group gives a singlet at δ_H 1.75, while the ^{13}C -NMR spectrum shows absorptions at δ_C 54.5 ($^1J_{PC}$ 117 Hz.) and 19.2 ppm ($^3J_{PC}$ 4.9 Hz.) assignable to the carbon bonded to phosphorus and the methyl group of the *E*-isomer²². Conversely, for **3a** the *Z*-isomer showed clearly different absorptions, namely a doublet at δ_H 3.58 ppm ($^2J_{PH}$ 19 Hz.) for the vinylic proton as well as a high-field signal for the methyl group at δ_H 2.07 ppm, while in the ^{13}C -NMR spectrum the absorption of methine carbon is shifted to higher field (δ_C 53.5) with a higher value of the phosphorus-carbon coupling constant ($^1J_{PC}$ 123 Hz.) relative to those of the *E*-isomer. Vicinal ^{13}C - ^{31}P coupling constant ($^3J_{PC}$ 15.4 Hz.) showed that the methyl group and phosphorus atom in the β -enamino compound **3a** are related *trans*²². In this context, it is worth noting that when chiral hydrazines such as *SAMP* and *RAMP* are used, only the *E*-isomer is obtained (see Table 1).



Scheme 2

Table 1. β -Enamino Phosphonium Salts **3** obtained.

| Compound | R ¹ | R ² | Yield (%) ^a | <i>E/Z</i> ratio ^b | m.p. (°C) |
|-----------|----------------|----------------|------------------------|-------------------------------|------------|
| 3a | Me | Me | 88 | 75 / 25 | 216-217(d) |
| 3b | Me | H | 91 | 0 / 100 | 134-135(d) |
| 3c | Ph | H | 70 | 37 / 63 | 189-190(d) |
| 3d | | H | 85 | 100 / 0 | 159-160(d) |
| 3e | | H | 79 | 100 / 0 | 159-160(d) |

^a Yield of isolated purified product. ^b *E/Z* ratio by ^{31}P -NMR assign.

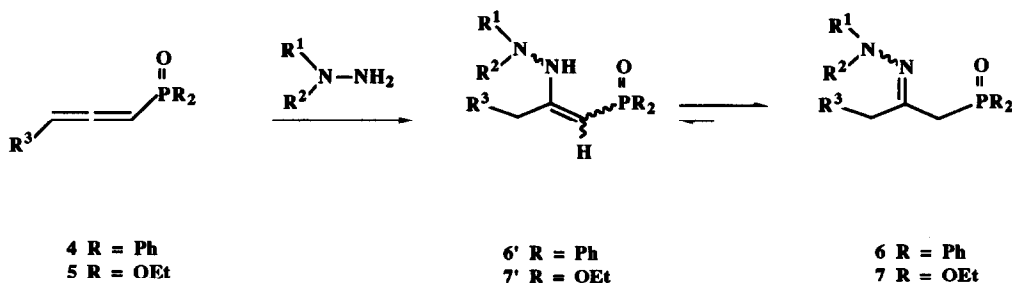
A mechanism that rationalizes the formation of **3** and that is consistent with the data is outlined in Scheme 2, in which hydrazines are caused to react with allenyltriphenylphosphonium bromide **2'** formed "in situ" from prop-2-ynyltriphenylphosphonium bromide and triethylamine²³. Hydrazines undergo a nucleophilic addition

with allene and give the Michael-type adduct **3'**. This material is readily rearranged under the reaction conditions to the thermodynamically more stable isomer, the β -enehydrazino phosphonium salts **3**.

These results prompted us to extend this reaction and to explore whether stable phosphorylated allenes with hydrazines showed a similar reaction pattern leading to new β -functionalized phosphorus compounds, that could be versatile key intermediates in carbon-carbon formation processes and in the synthesis of heterocycles.

Preparation of β -hydrazono phosphine oxides **6** and phosphonates **7**.

Simple addition of achiral and chiral hydrazines to allenes derived from phosphine oxides **4** in refluxing of chloroform (*TLC* control) leads, after crystallization of the crude reaction mixture, to the formation of β -hydrazono phosphine oxides **6** in excellent yield, instead of the tautomeric enehydrazino compounds **6'** such as have been obtained in the case of phosphonium salts **3** (Scheme 3, Table 2). Compounds **6** were characterized by their spectroscopic data, which indicate that they are isolated as a mixture of the *syn* and *anti* hydrazones **6**^{24,25}. Thus, the ³¹P-NMR spectrum of **6a** shows two absorptions at δ_p 27.8 and 29.5 ppm, in which the high-field chemical shift corresponds to the *syn*-isomer. Likewise, the ¹H- and ¹³C-NMR spectra show well resolved doublets for the methylene proton and carbon of **6a**; the proton absorptions of the *syn* isomer is



Scheme 3

Table 2. β -Hydrazono Phosphine Oxides **6** and Phosphonates **7**.

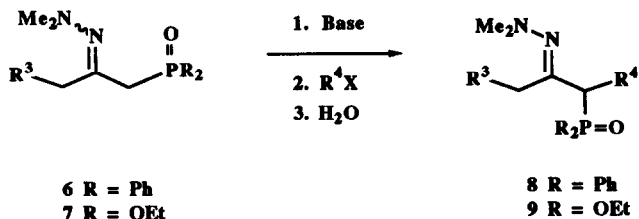
| Compound | R ¹ | R ² | R ³ | R | Yield (%) ^a | <i>syn/anti</i> ratio ^b | m.p. (°C) |
|-----------|----------------|----------------|----------------|-----|------------------------|------------------------------------|------------------|
| 6a | Me | Me | H | Ph | 89 | 24 / 76 | 107-108 |
| 6b | Me | Me | Me | Ph | 86 | 38 / 62 | 98-99 |
| 6c | | OMe | H | Ph | 90 | 15 / 85 | 91-92 |
| 6d | Me | H | H | Ph | 72 | 46 / 54 | 101-102 |
| 6e | Ph | H | H | Ph | 91 | 30 / 70 | 171-172 |
| 7a | Me | Me | H | OEt | 92 | 64 / 36 | oil ^c |
| 7b | Ph | H | H | OEt | 92 | 43 / 57 | oil ^c |

^a Yield of isolated purified product. ^b *Syn/anti* ratio by ³¹P-NMR assign. ^c Purified by flash chromatography.

shifted to a lower field δ_H 3.73 relative to that of the *anti*-isomer δ_H 3.34, while this latter isomer shows a downfield shift absorption for the methylene carbon δ_C 41.3 relative to that observed for the *syn*-isomer δ_C 33.7 ppm. Similarly, the allene derived from phosphonate ester **5** reacts with hydrazines and gives, after short flash column chromatography, β -functionalized phosphonates **7** in very high yield.

C-Alkylation of hydrazone anions derived from phosphine oxide and phosphonates.

Carbanions derived from hydrazones are especially useful in organic synthesis¹ taking into account: the reactivity of the intermediate azaallyl anions, the control over stereochemistry of electrophilic substitution step afforded by the nitrogen substituent, and the considerable control of regiochemistry which can be reached using hydrazones.



Scheme 4

Table 3. β -Functionalized Hydrazones **8** and **9** obtained.

| Compound | R ³ | R ⁴ | R | m.p.(°C) | Yield(%) ^a |
|-----------|----------------|-------------------------------------|-----|------------------|-----------------------|
| 8a | H | Me | Ph | 118-119 | 72 |
| 8b | H | CH ₂ -CH=CH ₂ | Ph | 124-125 | 80 |
| 8c | H | CH ₂ -COOCH ₃ | Ph | 84-85 | 83 |
| 8d | Me | Me | Ph | oil ^b | 79 |
| 9a | H | Me | OEt | oil ^b | 71 |

^a Yield of isolated compounds from **6** and **7**. ^b Purified by flash-chromatography.

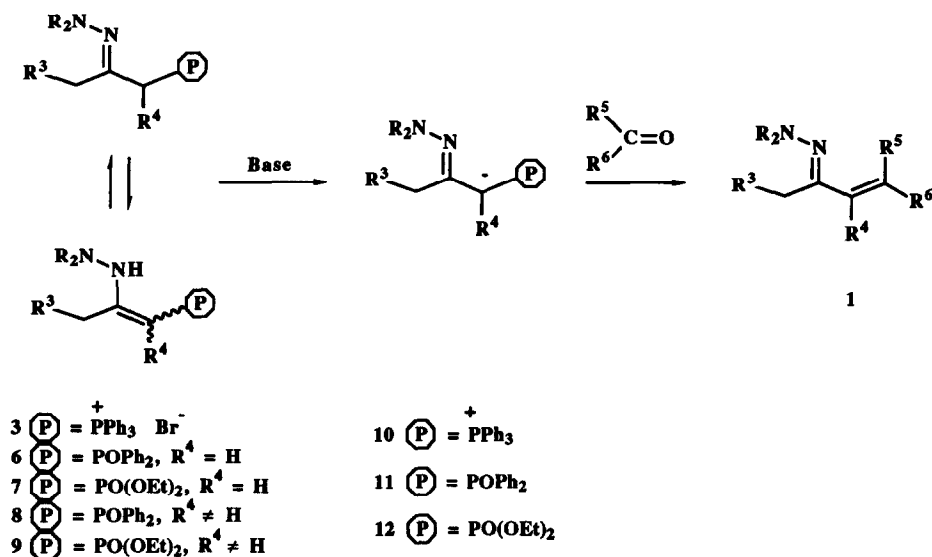
In our case, moreover, the presence of an anion stabilizing group such as phosphine oxide **6** or phosphonate **7** could control the deprotonation at the internal less-substituted carbon. Thus, when β -phosphorylated hydrazones **6** and **7** were treated with lithium diisopropylamide (*LDA*) followed by addition of alkyl halides and aqueous work-up, substituted compounds **8** and **9** were obtained (Scheme 4, Table 3).

Olefination reaction of β -functionalized phosphorus derivatives **3**, **6** and **7**.

As we had proposed in Scheme 1, phosphorus compounds could be suitable to efficiently achieve the homologation of hydrazones into their vinylogous compounds. Thus, achiral and chiral phosphonium salts **3** were treated with a base followed by Wittig reaction of the phosphorane **10** with aliphatic, heteroaromatic and aromatic aldehydes (Scheme 5) leading to 1-azadienes **1** with high *E* stereoselectivity of the carbon-carbon

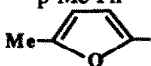
double bond in excellent yield (Table 4), after aqueous work-up and flash- chromatography. *n*-Buthyl lithium was the initial base chosen for the “*in situ*” generation of phosphorane **10**, although owing to the partially stabilised nature of this phosphorane, a weaker base such as potassium carbonate would suffice. The use of this base requires no special precautions and provides excellent yields (Scheme 5, Table 4). The structures of products **1** were ascertained on the basis of their spectroscopic data, which indicate that they are isolated as the *syn*- and *anti*-isomers. Vicinal 3HH coupling constants in the range of 16-17 Hz. between the vinylic protons of **1** are consistent with the *E* configuration of the carbon-carbon double bond. Therefore, this procedure is highly stereoselective affording exclusively the *E* stereoisomer.

Wittig reaction of phosphoranes **10** generated “*in situ*” from phosphonium salts **3** and simple ketones fails, probably due to the partially stabilized character of these phosphoranes **10**. With these results in mind, we attempted to extend this methodology for two carbon homologation of hydrazones, by using, instead of phosphonium salts **3**, the corresponding β -functionalized phosphines oxides **6** and phosphonates **7**. Thus, metalation of β -hydrazone phosphine oxides **6** and phosphonates **7** with methyl lithium or *LDA* followed by the addition of aldehydes and ketones led to the formation of α,β -unsaturated hydrazones **1**. While these olefination reactions by using **6** and **7** with aldehydes gave similar yields (Table 4) to that obtained in the case of phosphoranes **10**, these phosphorylated substrates **6** and **7** are especially useful for the elongation of ketones. It is noteworthy that the preparation of *C*- α -substituted hydrazones **1q**, **1r** and **1s** does not require the isolation and purification of the phosphine oxides **8a** and **8b** or the phosphonate **9a**. They can be obtained in a “*one pot*” reaction from **6** and **7** when these compounds are directly metallated in *THF* with subsequent addition of alkyl halide, a second equivalent of base, aldehydes and aqueous work-up, respectively.

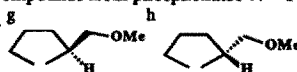


Scheme 5

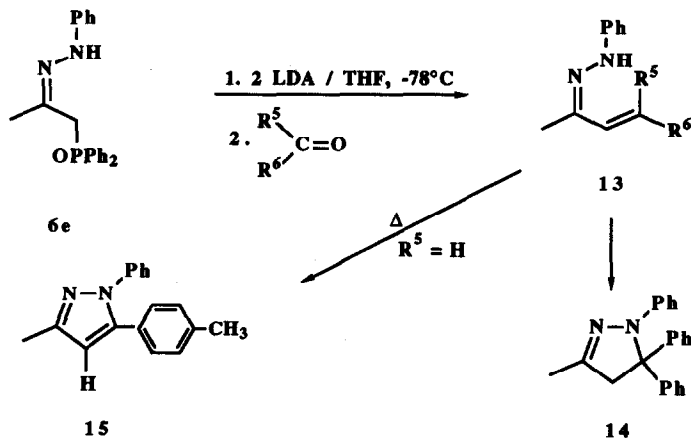
Table 4. α,β -Unsaturated Hydrazones 1 obtained.

| Compound ^a | R ₂ | R ³ | R ⁴ | R ⁵ | R ⁶ | Yield.(%) |
|-----------------------|-----------------|----------------|---------------------------------------|-------------------------------------|---|---|
| 1a | Me ₂ | H | H | H | p-O ₂ N-Ph | 90 ^b |
| 1b | Me ₂ | H | H | H | Me-CH=CH | 84 ^b |
| 1c | Me ₂ | H | H | H | Ph | 82 ^b |
| 1d | Me ₂ | H | H | H | p-Me-Ph | 81 ^b 77 ^c 72 ^d |
| 1e | Me ₂ | H | H | H |  | 81 ^b |
| 1f | Me ₂ | H | H | H | Ph-CH ₂ -CH ₂ | 91 ^b |
| 1g | Me ₂ | H | H | H | (CH ₃) ₂ -CH-CH ₂ | 82 ^b |
| 1h | g | H | H | H | p-Me-Ph | 88 ^b |
| 1i | h | H | H | H | p-Me-Ph | 75 ^b |
| 1j | Me ₂ | H | H | H | EtO ₂ C | 80 ^b |
| 1k | Me ₂ | Me | H | H | p-Me-Ph | 85 ^c |
| 1l | Me ₂ | H | H | Ph | Ph | 80 ^c |
| 1m | Me ₂ | H | H | - (CH ₂) ₅ - | | 86 ^c |
| 1n | Me ₂ | H | H | Me | ⁱ Bu | 68 ^c |
| 1o | g | H | H | H | Ph | 93 ^c |
| 1p | Me ₂ | H | H | H | p-MeO-Ph | 74 ^d |
| 1q | Me ₂ | H | Me | H | p-Me-Ph | 92 ^e 66 ^c |
| 1r | Me ₂ | H | CH ₂ =CH-CH ₂ - | H | p-Me-Ph | 72 ^e 57 ^c |
| 1s | Me ₂ | H | Me | H | p-MeO-Ph | 76 ^f 60 ^d |

^a Purified by flash-chromatography. ^b Yield of isolated compounds from phosphoranes 10. ^c Yield of isolated compounds from phosphine oxides 6. ^d Yield of isolated compounds from phosphonates 7. ^e Yield of isolated compounds from 8. ^f Yield of isolated compounds from 9. ^g



Finally, this strategy used for the preparation of α,β -unsaturated hydrazones can also be applied for five membered heterocycle formation when *N*-aryl hydrazones 6e are used. In such substances, one or both nitrogen atoms of functionalized hydrazones can be used for the cyclisation. Acyclic α,β -unsaturated compound 13 is formed by deprotonation of both the N-H and the more substituted α -carbon of the starting *N*-phenyl hydrazone 6e with two equivalents of a strong base like *LDA* followed by addition of *p*-tolyl aldehyde and work-up. Heating 13 at 100°C in toluene causes intramolecular Michael addition and gives pyrazole 15. However, when diphenyl ketone reacts with the dianion from phosphorylated hydrazone 6e, 1-azadiene 13 is not isolated and pyrazoline 14 is obtained directly instead (Scheme 6).



Scheme 6

In conclusion, β -functionalized phosphonium salts 3, phosphine oxides 6 and phosphonates 7 described here are easily synthesized intermediates which can be used for an effective, versatile and high yielding synthesis of α,β -unsaturated hydrazones and five membered heterocycles.

ACKNOWLEDGEMENTS

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EXPERIMENTAL SECTION

General. Melting points were determined with a Buchi SPM-20 apparatus and are uncorrected. Analytical TLC was performed on 0.25mm silica gel plates (Merck). Visualization was accomplished by UV light and iodine. Solvents for extraction and chromatography were technical grade and distilled from the indicated drying agents: CH_2Cl_2 (P_2O_5); Hexane and diethyl ether (sodium benzophenone ketyl); ethyl acetate (K_2CO_3). All solvents used in reactions were freshly distilled from appropriate drying agents before use: THF (sodium benzophenone ketyl); DMF (CaH_2); CHCl_3 (P_2O_5). All other reagents were recrystallized or distilled as necessary. Column (flash) chromatography was carried out on silica gel (Merck, 70-230 mesh). Mass spectra were obtained on a Hewlett Packard 5890 spectrometer. Infrared spectra were taken on a Nicolet IRFT Magna 550 spectrometer. $^1\text{H-NMR}$ spectra were recorded on a Bruker 250 MHz spectrometer using tetramethylsilane (0.00 ppm) or chloroform (7.26 ppm) as an internal reference in CDCl_3 solutions. $^{13}\text{C-NMR}$ spectra were recorded at 75 MHz with chloroform (77.0 ppm) as an internal reference in CDCl_3 solutions. $^{31}\text{P-NMR}$ spectra were recorded at 120 MHz with 85% phosphoric acid as an external reference. Elemental analyses were performed in a Perkin Elmer Model 240 instrument. Chemical shifts are given in ppm (δ); multiplicities are indicated by s (singlet), d (doublet), dd (double-doublet), t (triplet), q (quadruplet) or m (multiplet). Coupling constants, J , are reported in hertz. Infrared spectra (IR) were obtained as neat liquids, or as solids in KBr . Peaks are reported in cm^{-1} . Mass spectra (EI) were obtained with a ionization voltage of 70 eV. Data are reported in the form m/z (intensity relative to base = 100). All reactions were performed in oven (125 $^\circ\text{C}$) or flame-dried glassware under an inert atmosphere of dry N_2 .

The following compounds were prepared by literature methods: 1,2-propadienyldiphenylphosphine oxide²⁶, 1,2-butadienyldiphenylphosphine oxide²⁶ and diethyl 1,2-propadienyldiphenylphosphonate²⁶.

General Procedure for the Preparation of the β -Hydrazinoprop-1-enylphosphonium Bromides 3. A dry flask, 100-mL, 3-necked, fitted with a reflux condenser, gas inlet, dropping funnel, and magnetic stirrer, was charged 1.9 g (5

mmol) of popargyltriphenylphosphonium bromide **2**, 0.83 mL (6 mmol) of triethylamine and 30 mL of $CHCl_3$. A solution (5 mmol) of hydrazine and 20 mL of $CHCl_3$ was added over 10 min. The mixture was stirred and refluxed until TLC indicated the disappearance of the phosphonium salt (1 day to 3 days). The mixture was diluted with 50 mL water and extracted with CH_2Cl_2 (3 x 25 mL). The CH_2Cl_2 layers were washed with water. The combined organic layers were dried over $MgSO_4$, filtered, and concentrated the crude product was purified by recrystallization ($CHCl_3$ / ethyl acetate).

Z- and E- β -dimethylhydrazino prop-1-enylphosphonium bromide (3a). 1936 mg (88 %) of **3a** as a yellow solid. Data for **3a**: mp 216-217 °C (dec.); 1H -NMR (250 MHz) 1.75 and 2.07 (s, 3H, E- and Z-CH₃), 2.32 and 2.72 (s, 6H, Z- and E-CH₃N), 3.58 (d, 1H, $^2J_{PH}$ = 19 Hz, Z-CH), 4.61 (d, 1H, $^2J_{PH}$ = 18 Hz, E-CH), 7.39-7.75 (m, 15H, arom), 9.72 and 10.01 (s, 1H, Z- and E-NH); ^{13}C -NMR (75 MHz) 19.2 (d, $^3J_{PC}$ = 4.9 Hz, E-CH₃), 22.0 (d, $^3J_{PC}$ = 15.4 Hz, Z-CH₃), 45.8 and 46.0 (E- and Z-CH₃N), 53.5 (d, $^1J_{PC}$ = 123 Hz, Z-CH), 54.5 (d, $^1J_{PC}$ = 117 Hz, E-CH), 122.8-133.8 (C-arom), 164.2 and 164.4 (E- and Z-C-N); ^{31}P -NMR (120 MHz) 11.8 (Z-isomer), 17.4 (E-isomer); IR (KBr) 3402, 3117, 2968, 2950, 1099; MS (70 eV) 360 (M⁺-HBr, 5). Anal. Calcd for C₂₃H₂₆N₂PBr (440.10): C, 62.71; H, 5.95; N, 6.36. Found: C, 62.52; H, 5.97; N, 6.34.

Z- β -methylhydrazino prop-1-enylphosphonium bromide (3b). 1939 mg (91 %) of **3b** as a yellow solid. Data for **3b**: mp 134-135 °C (dec.); 1H -NMR (250 MHz) 2.16 (s, 3H, CH₃), 3.38 (s, 3H, CH₃N), 3.51 (d, 1H, $^2J_{PH}$ = 22.4 Hz, CH), 4.40 (s, 1H, NH), 7.39-7.63 (m, 16H, arom and NH); ^{13}C -NMR (75 MHz) 23.3 (d, $^3J_{PC}$ = 14.6 Hz, CH₃), 42.8 (CH₃N), 58.7 (d, $^1J_{PC}$ = 120.7 Hz, CH), 128.4-133.8 (C-arom), 166.3 (C-N); ^{31}P -NMR (120 MHz) 14.1; IR (KBr) 3435, 3203, 3111, 1656, 1550, 1430, 1101; MS (70 eV) 346 (M⁺-HBr, 6). Anal. Calcd. for C₂₂H₂₄N₂PBr (426.09): C, 61.96; H, 5.68; N, 6.57. Found: C, 61.72; H, 5.69; N, 6.59.

Z- and E- β -phenylhydrazino prop-1-enylphosphonium bromide (3c). 1708 mg (70 %) of **3c** as a yellow solid. Data for **3c**: mp 188-190 °C (dec.); 1H -NMR (250 MHz) 1.66 and 2.03 (s, 3H, E- and Z-CH₃), 3.90 (d, 1H, $^2J_{PH}$ = 21.4 Hz, Z-CH), 4.81 (s, 1H, NH), 5.00 (d, 1H, $^2J_{PH}$ = 14.3 Hz, E-CH), 5.35 (s, 1H, NH), 7.19-7.73 (m, 20H, arom); ^{13}C -NMR (75 MHz) 21.5 (d, $^3J_{PC}$ = 5.4 Hz, E-CH₃), 24.3 (d, $^3J_{PC}$ = 14 Hz, Z-CH₃), 63.0 (d, $^1J_{PC}$ = 118.4 Hz, CH), 122.5-144.4 (C-arom), 165.7 (C-N); ^{31}P -NMR (120 MHz) 15.3 (Z-isomer), 16.9 (E-isomer); IR (KBr) 3237, 3059, 1532, 1440, 1104; MS (70 eV) 408 (M⁺-HBr, 12). Anal. Calcd for C₂₇H₂₆N₂PBr (488.10): C, 66.38; H, 5.37; N, 5.74. Found: C, 66.56; H, 5.35; N, 5.72.

E- β -N-(S)-(-)-2-methoxymethylpyrrolidineamino prop-1-enylphosphonium bromide (3d). 2168 mg (85 %) of **3d** as a yellow solid. Data for **3d**: mp 159-160 °C (dec.); 1H -NMR (250 MHz) 1.66-2.17 (m, 6H, ring CH₂), 1.75 (s, 3H, CH₃), 3.11-3.68 (m, 3H, CH₂O and CH), 3.16 (s, 3H, OCH₃), 4.84 (d, 1H, $^2J_{PH}$ = 18.2 Hz, CH), 7.41-7.67 (m, 15H, arom), 9.68 (s, 1H, NH); ^{13}C -NMR (75 MHz) 18.8 (d, $^3J_{PC}$ = 5.1 Hz, CH₃), 20.7 (ring CH₂), 24.7 (ring CH₂), 52.9 (d, $^1J_{PC}$ = 60.7 Hz, CH), 53.9 (ring CH₂), 58.3 (OCH₃), 64.1 (CH), 72.9 (CH₂O), 122.5-133.1 (C-arom), 164.5 (C-N); ^{31}P -NMR (120 MHz) 17.5; IR (KBr) 3510, 3370, 3180, 3040, 1567, 1442; MS (70 eV) 430 (M⁺-HBr, 3). Anal. Calcd. for C₂₇H₃₂N₂OPBr (510.14): C, 63.51; H, 6.32; N, 5.49. Found: C, 63.65; H, 6.31; N, 5.47.

E- β -N-(R)-(+)-2-methoxymethylpyrrolidineamino prop-1-enylphosphonium bromide (3e). 2015 mg (79 %) of **3e** as a yellow solid. Data for **3e**: mp 159-160 °C (dec.); 1H -NMR (250 MHz) 1.69-2.20 (m, 6H, ring CH₂), 1.80 (s, 3H, CH₃), 3.14-3.51 (m, 3H, CH₂O and CH), 3.19 (s, 3H, OCH₃), 4.87 (d, 1H, $^2J_{PH}$ = 18.3 Hz, CH), 7.56-7.84 (m, 15H, arom), 9.79 (s, 1H, NH); ^{13}C -NMR (75 MHz) 19.5 (d, $^3J_{PC}$ = 5.2 Hz, CH₃), 21.4 (ring CH₂), 25.3 (ring CH₂), 52.7 (d, $^1J_{PC}$ = 61.5 Hz, CH), 52.7 (ring CH₂), 54.6 (OCH₃), 59.0 (CH), 73.5 (CH₂O), 123.2-133.8 (C-arom), 165.2 (C-N); ^{31}P -NMR (120 MHz) 17.6; IR (KBr) 3500, 3370, 3178, 3037, 1565, 1440; MS (70 eV) 431 (M⁺-Br, 100). Anal. Calcd. for C₂₇H₃₂N₂OPBr (510.14): C, 63.51; H, 6.32; N, 5.49. Found: C, 63.66; H, 6.30; N, 5.47.

General Procedure for the Preparation of the β -Hydrazonoalkyldiphenylphosphine oxides **6, and diethyl β -Hydrazonopropylphosphonates **7**.** A dry flask, 100-mL, 3-necked, fitted with a reflux condenser, gas inlet, dropping funnel, and magnetic stirrer, was charged 1.2 g (5 mmol) of allenediphenylphosphine oxide **4** (R³ = H), or 1.27 g (5 mmol) of 1,2-butadienyldiphenylphosphine oxide **4** (R³ = CH₃), or 0.88 g (5 mmol) of diethyl 1,2-propadienyldiphenylphosphonate **5** (R³ = H), and 30 mL of $CHCl_3$. A solution (5 mmol) of hydrazine and 20 mL of $CHCl_3$ was added over 10 min. The mixture was stirred and refluxed until TLC indicated the disappearance of the phosphine oxide or phosphonate (1 day to 3 days). The mixture was concentrated and the crude product was purified by recrystallization (hexane / ethyl acetate).

Syn- and anti- β -N,N-dimethylhydrazonopropylidiphenylphosphine oxide (6a). 1336 mg (89 %) of **6a** as a white solid. Data for **6a**: mp 107-108 °C; 1H -NMR (250 MHz) 2.08 and 2.09 (s, 3H, anti- and syn-CH₃), 2.18 and 2.22 (s, 6H, anti- and syn-CH₃N), 3.34 (d, 2H, $^2J_{PH}$ = 14.3 Hz, anti-CH₂), 3.73 (d, 2H, $^2J_{PH}$ = 14.9 Hz, syn-CH₂), 7.27-7.77 (m, 10H, arom); ^{13}C -NMR (75 MHz) 18.6 and 24.2 (anti- and syn-CH₃), 33.7 (d, $^1J_{PC}$ = 64 Hz, syn-CH₂), 41.3 (d, $^1J_{PC}$ = 64.5 Hz, anti-CH₂), 46.3 and 46.6 (anti- and syn-CH₃N), 127.9-133.1 (C-arom), 159.9 and 160.0 (anti- and syn-C=N); ^{31}P -NMR (120 MHz)

27.8 and 29.5 (*syn*- and *anti*-isomers); IR (KBr) 2943, 2881, 1438, 1189; MS (70 eV) 300 (M^+ , 6). Anal. Calcd. for $C_{17}H_{21}N_2OP$ (300.14): C, 67.97; H, 7.05; N, 9.33. Found: C, 68.14; H, 7.02; N, 9.31.

Syn- and anti- β -N,N-dimethylhydrazonobutyldiphenylphosphine oxide (6b). 1351 mg (86 %) of 6b as a white solid. Data for 6b: mp 98-99 °C; 1H -NMR (250 MHz) 1.02-1.14 (m, 3H, *syn*- and *anti*-CH₃), 2.15 and 2.21 (s, 6H, *syn*- and *anti*-CH₃N), 2.43-2.45 and 2.46-2.59 (m, 2H, *anti*- and *syn*-CH₂(Et)), 3.37 (d, 2H, $^2J_{PH} = 14.5$ Hz, *syn*-CH₂), 3.71 (d, $^2J_{PH} = 14.9$ Hz, *anti*-CH₂), 7.26-7.82 (m, 10H, arom); ^{13}C -NMR (75 MHz) 11.2 and 11.3 (*anti*- and *syn*-CH₃), 24.6 and 30.8 (*anti*- and *syn*-CH₂(Et)), 32.3 (d, $^1J_{PC} = 64.5$ Hz, *syn*-CH₂), 38.2 (d, $^1J_{PC} = 64.8$ Hz, *anti*-CH₂), 46.9 and 47.3 (*syn*- and *anti*-CH₃N), 128.3-137.8 (C-arom), 165.8 and 165.9 (*syn*- and *anti*-C=N); ^{31}P -NMR (120 MHz) 27.9 and 29.8 (*anti*- and *syn*-isomers); IR (KBr) 1704, 1617, 1435, 1181; MS (70 eV) 314 (M^+ , 4). Anal. Calcd. for $C_{18}H_{23}N_2OP$ (314.16): C, 68.76; H, 7.38; N, 8.92. Found: C, 68.57; H, 7.39; N, 8.94.

Syn- and anti- β -N-(S)-(-)-2-methoxymethylpyrrolidineimino propyldiphenylphosphine oxide (6c). 1666 mg (90 %) of 6c as a white solid. Data for 6c: mp 91-92 °C; 1H -NMR (250 MHz) 1.37-1.88 (m, 6H, ring CH₂), 1.91 and 1.92 (s, 3H, *anti*- and *syn*-CH₃), 2.85-3.21 (m, 3H, CH₂O and CH), 3.12 (s, 3H, OCH₃), 3.27 (d, 2H, $^2J_{PH} = 14.5$ Hz, *syn*-CH₂), 3.50 (d, 2H, $^2J_{PH} = 14.7$ Hz, *anti*-CH₂), 7.27-7.74 (m, 10H, arom); ^{13}C -NMR (75 MHz) 20.5 and 24.8 (*anti*- and *syn*-CH₃), 22.3 and 22.5 (ring *syn*- and *anti*-CH₂), 26.5 and 26.8 (ring *syn*- and *anti*-CH₂), 34.8 (d, $^1J_{PC} = 35.6$ Hz, *syn*-CH₂), 41.8 (d, $^1J_{PC} = 35$ Hz, *anti*-CH₂), 54.3 and 54.4 (ring *anti*- and *syn*-CH₂), 59.2 (*syn*- and *anti*-OCH₃), 66.1 and 66.5 (*syn*- and *anti*-CH), 74.9 and 75.4 (*syn*- and *anti*-CH₂O), 128.3-133.4 (C-arom), 156.7 and 161.4 (*anti*- and *syn*-C=N); ^{31}P -NMR (120 MHz) 27.4 and 29.7 (*syn*- and *anti*-isomers); IR (KBr) 1636, 1591, 1553, 1122; MS (70 eV) 370 (M^+ , 100). Anal. Calcd. for $C_{21}H_{27}N_2O_2P$ (370.18): C, 68.07; H, 7.35; N, 7.57. Found: C, 68.26; H, 7.37; N, 7.54.

Syn- and anti- β -N-methylhydrazonopropyldiphenylphosphine oxide (6d). 1030 mg (72 %) of 6d as a white solid. Data for 6d: mp 101-102 °C; 1H -NMR (250 MHz) 1.54 and 1.82 (s, 3H, *anti*- and *syn*-CH₃), 2.76 and 2.90 (s, 3H, *syn*- and *anti*-CH₃N), 3.34 (d, 2H, $^2J_{PH} = 14$ Hz, *syn*-CH₂), 3.44 (d, 2H, $^2J_{PH} = 14.5$ Hz, *anti*-CH₂), 4.60 (s, 1H, NH), 7.26-7.81 (m, 10H, arom); ^{13}C -NMR (75 MHz) 15.9 and 25.3 (*anti*- and *syn*-CH₃), 34.9 (d, $^1J_{PC} = 64.5$ Hz, *syn*-CH₂), 37.9 and 38.1 (*anti*- and *syn*-CH₃N), 41.4 (d, $^1J_{PC} = 66.4$ Hz, *anti*-CH₂), 128.4-133.5 (C-arom), 140.0 and 141.2 (*syn*- and *anti*-C=N); ^{31}P -NMR (120 MHz) 29.7 and 31.2 (*anti*- and *syn*-isomers); IR (KBr) 3263, 2939, 1664, 1435, 1177; MS (70 eV) 286 (M^+ , 32). Anal. Calcd. for $C_{16}H_{19}N_2OP$ (286.12): C, 67.10; H, 6.69; N, 9.79. Found: C, 66.96; H, 6.72; N, 9.76.

Syn- and anti- β -N-phenylhydrazonopropyldiphenylphosphine oxide (6e). 1584 mg (91 %) of 6e as a white solid. Data for 6e: mp 171-172 °C; 1H -NMR (250 MHz) 1.58 and 1.91 (s, 3H, *anti*- and *syn*-CH₃), 3.38 (d, 2H, $^2J_{PH} = 14$ Hz, *syn*-CH₂), 3.42 (d, 2H, $^2J_{PH} = 14.7$ Hz, *anti*-CH₂), 6.73-7.79 (m, 15H, arom), 9.67 (s, 1H, NH); ^{13}C -NMR (75 MHz) 16.3 and 25.7 (*anti*- and *syn*-CH₃), 35.9 (d, $^1J_{PC} = 64.6$ Hz, *syn*-CH₂), 41.2 (d, $^1J_{PC} = 66.2$ Hz, *anti*-CH₂), 112.9-132.5 (C-arom), 145.2 and 146.9 (*anti*- and *syn*-C=N); ^{31}P -NMR (120 MHz) 30.2 and 31.7 (*syn*- and *anti*-isomers); IR (KBr) 3256, 1604, 1506, 1439, 1150; MS (70 eV) 348 (M^+ , 100). Anal. Calcd. for $C_{21}H_{21}N_2OP$ (348.14): C, 72.38; H, 6.08; N, 8.04. Found: C, 72.54; H, 6.06; N, 8.06.

Syn- and anti-diethyl β -N,N-dimethylhydrazonopropylphosphonate (7a). The crude product was purified by flash-chromatography (hexane / diethyl ether, 1 / 1) to afford 1086 mg (92 %) of 7a as a yellow oil ($R_f = 0.13$, ethyl acetate). Data for 7a: 1H -NMR (250 MHz) 1.16-1.23 (m, 6H, *syn*- and *anti*-CH₃(Et)), 1.74 and 1.81 (s, 3H, *anti*- and *syn*-CH₃), 2.12 and 2.16 (s, 6H, *anti*- and *syn*-CH₃N), 2.53 (d, 2H, $^2J_{PH} = 22.3$ Hz, *syn*-CH₂), 3.02 (d, 2H, $^2J_{PH} = 23.5$ Hz, *anti*-CH₂), 3.96-4.02 (m, 4H, *syn*- and *anti*-CH₂(Et)); ^{13}C -NMR (75 MHz) 16.0 and 16.1 (*anti*- and *syn*-CH₃(Et)), 17.8 and 23.6 (*anti*- and *syn*-CH₃), 29.5 (d, $^1J_{PC} = 136$ Hz, *syn*-CH₂), 36.9 (d, $^1J_{PC} = 134.3$ Hz, *anti*-CH₂), 46.6 and 46.9 (*anti*- and *syn*-CH₃N), 61.6 and 61.7 (*anti*- and *syn*-CH₂(Et)), 159.0 and 160.2 (*anti*- and *syn*-C=N); ^{31}P -NMR (120 MHz) 24.1 and 25.2 (*anti*- and *syn*-isomers); IR 1644, 1453, 1256, 1019; MS (70 eV) 236 (M^+ , 37). Anal. Calcd. for $C_9H_{21}N_2O_3P$ (236.13): C, 45.74; H, 8.96; N, 11.86. Found: C, 45.63; H, 8.99; N, 11.82.

Syn- and anti-diethyl β -N-phenylhydrazonopropylphosphonate (7b). The crude product was purified by flash-chromatography (hexane / diethyl ether, 1 / 1) to afford 1307 mg (92 %) of 7b as a yellow oil ($R_f = 0.51$, ethyl acetate). Data for 7b: 1H -NMR (250 MHz) 1.26-1.38 (m, 6H, *syn*- and *anti*-CH₃(Et)), 1.97 and 2.10 (s, 3H, *anti*- and *syn*-CH₃), 2.91 (d, 2H, $^2J_{PH} = 21.7$ Hz, *syn*-CH₂), 2.93 (d, 2H, $^2J_{PH} = 23$ Hz, *anti*-CH₂), 4.05-4.19 (m, 4H, *syn*- and *anti*-CH₂(Et)), 6.76-7.34 (m 5H, arom), 8.50 (s, 1H, NH); ^{13}C -NMR (75 MHz) 15.5 and 25.1 (*anti*- and *syn*-CH₃), 16.3 and 16.4 (*syn*- and *anti*-CH₃(Et)), 30.5 (d, $^1J_{PC} = 135.9$ Hz, *syn*-CH₂), 36.7 (d, $^1J_{PC} = 136$ Hz, *anti*-CH₂), 62.0 and 62.7 (*anti*- and *syn*-CH₂(Et)), 112.1-129.0 (C-arom); 145.6 and 146.5 (*anti*- and *syn*-C=N); ^{31}P -NMR (120 MHz) 24.7 and 25.8 (*anti*- and *syn*-isomers); IR 3281, 1608, 1498, 1241, 1028; MS (70 eV) 284 (M^+ , 100). Anal. Calcd. for $C_{13}H_{21}N_2O_3P$ (284.13): C, 54.90; H, 7.45; N, 9.86. Found: C, 54.71; H, 7.43; N, 9.88.

General Procedure for the C-alkylation of β -Hydrazonoalkyldiphenylphosphine oxides 6 and diethyl β -hydrazonopropylphosphonates 7. A dry flask, 100-mL, 2-necked, fitted with a dropping funnel, gas inlet, and magnetic stirrer, was charged 5 mmol of lithium diisopropylamide (LDA) and 45 mL of THF. The temperature was allowed to descend to -78°C and a solution 1.5 g (5 mmol) of β -*N,N*-dimethylhydrazonopropylidiphenylphosphine oxide 6a, or 1.6 g (5 mmol) of β -*N,N*-dimethylhydrazonobutylidiphenylphosphine oxide 6b, or 1.2 g (5 mmol) of diethyl β -*N,N*-dimethylhydrazonopropylphosphonate 7a in 40 mL of THF was then added. The mixture was allowed to stir for 1 h. A solution 5 mmol of alkyl halide in 10 mL of THF was added at -78°C . The mixture was stirred until TLC indicated the disappearance of compounds 6a, 6b or 7a (3 h to 2 days). The mixture was diluted with 50 mL water and extracted with CH_2Cl_2 . The CH_2Cl_2 layers were washed with water. The organic layers were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by recrystallization from CH_2Cl_2 / hexane or by flash-chromatography (hexane / diethyl ether, 1 / 1).

α -Methyl- β -*N,N*-dimethylhydrazonopropylidiphenylphosphine oxide (8a). 1131 mg (72 %) of 8a as white solid. Data for 8a: mp 118-119 $^\circ\text{C}$; $^1\text{H-NMR}$ (250 MHz) 1.37 (dd, 3H, $^3J_{\text{HH}} = 7.3$ Hz, $^3J_{\text{PH}} = 15.9$ Hz, CH₃), 2.00 (s, 3H, CH₃), 2.12 (s, 6H, CH₃N), 3.46-3.52 (m, 1H, CH), 7.26-7.86 (m, 10H, arom); $^{13}\text{C-NMR}$ (75 MHz) 11.2 (CH₃), 15.3 (CH₃), 43.9 (d, $^1J_{\text{PC}} = 65.7$ Hz, CH), 46.5 (CH₃N), 128.1-131.6 (C-arom), 165.6 (C=N); $^{31}\text{P-NMR}$ (120 MHz) 32.2; IR (KBr) 2973, 2848, 1440, 1203; MS (70 eV) 314 (M^+ , 9). Anal. Calcd for C₁₈H₂₃N₂OP (314.16): C, 68.76; H, 7.38; N, 8.91. Found: C, 68.54; H, 7.40; N, 8.88.

α -Allyl- β -*N,N*-dimethylhydrazonopropylidiphenylphosphine oxide (8b). 1361 mg (80 %) of 8b as white solid. Data for 8b: mp 124-125 $^\circ\text{C}$; $^1\text{H-NMR}$ (250 MHz) 2.01 (s, 3H, CH₃), 2.15 (s, 6H, CH₃N), 2.23-2.46 (m, 1H, CH₂), 2.73-2.87 (m, 1H, CH₂), 3.46-3.56 (m, 1H, CH), 4.94-5.06 (m, 2H, H₂C=), 5.56-5.69 (m, 1H, =CH), 7.26-7.91 (m, 10H, arom); $^{13}\text{C-NMR}$ (75 MHz) 15.9 (CH₃), 29.6 (CH₂), 46.6 (CH₃N), 49.3 (d, $^1J_{\text{PC}} = 64.5$ Hz, CH), 116.9 (H₂C=), 128.1-131.7 (C-arom), 134.5 (d, $^3J_{\text{PC}} = 14.5$ Hz, =CH), 164.1 (C=N); $^{31}\text{P-NMR}$ (120 MHz) 31.3; IR (KBr) 2967, 2861, 1440, 1177; MS (70 eV) 340 (M^+ , 2). Anal. Calcd. for C₂₀H₂₅N₂O₂P (340.17): C, 70.55; H, 7.41; N, 8.23. Found: C, 70.28; H, 7.43; N, 8.21.

α -Methoxycarbonylmethyl- β -*N,N*-dimethylhydrazonopropylidiphenylphosphine oxide (8c). 1544 mg (83 %) of 8c as white solid. Data for 8c: mp 84-85 $^\circ\text{C}$; $^1\text{H-NMR}$ (250 MHz) 1.98 (s, 3H, CH₃), 2.19 (s, 6H, CH₃N), 2.57-2.68 (m, 1H, CH₂), 3.04-3.49 (m, 1H, CH₂), 3.58 (s, 3H, OCH₃), 3.62-3.91 (m, 1H, CH), 7.26-7.86 (m, 10H, arom); $^{13}\text{C-NMR}$ (75 MHz) 17.0 (CH₃), 31.5 (CH₂), 46.5 (d, $^1J_{\text{PC}} = 64.5$ Hz, CH), 46.5 (CH₃N), 127.6-132.1 (C-arom), 162.7 (C=N), 171.4 (d, $^3J_{\text{PC}} = 17.8$ Hz, C=O); $^{31}\text{P-NMR}$ (120 MHz) 31.1; IR (KBr) 2973, 2861, 1756, 1446, 1190; MS (70 eV) 372 (M^+ , 9). Anal. Calcd. for C₂₀H₂₅N₂O₃P (372.16): C, 64.49; H, 6.77; N, 7.52. Found: C, 64.62; H, 6.75; N, 7.50.

α -Methyl- β -*N,N*-dimethylhydrazonobutylidiphenylphosphine oxide (8d). 1296 mg (79 %) of 8d as a yellow oil ($R_f = 0.51$, acetone). Data for 8d: $^1\text{H-NMR}$ (250 MHz) 0.84-1.05 (m, 3H, CH₃(Et)), 1.29-1.44 (m, 3H, CH₃), 2.26 (s, 6H, CH₃N), 2.36-2.55 (m, 2H, CH₂), 3.47-3.62 (m, 1H, CH), 7.26-7.79 (m, 10H, arom); $^{13}\text{C-NMR}$ (75 MHz) 7.3 (CH₃), 11.2 (CH₃), 26.7 (CH₂), 43.1 (d, $^1J_{\text{PC}} = 66$ Hz, CH), 47.3 (CH₃N), 128.0-132.9 (C-arom), 169.9 (C=N); $^{31}\text{P-NMR}$ (120 MHz) 31.3; IR (KBr) 1702, 1438, 1187, 1118; MS (70 eV) 328 (M^+ , 14). Anal. Calcd for C₁₉H₂₅N₂O₂P (328.17): C, 69.48; H, 7.68; N, 8.53. Found: C, 69.66; H, 7.65; N, 8.51.

α -Methyl diethyl β -*N,N*-dimethylhydrazonopropylphosphonate (9a). 888 mg (71 %) of 9a as a yellow oil ($R_f = 0.49$, acetone). Data for 9a: $^1\text{H-NMR}$ (250 MHz) 1.17-1.36 (m, 9H, CH₃ and CH₃(Et)), 1.98 (s, 3H, CH₃), 2.39 (s, 6H, CH₃N), 2.83-2.94 (m, 1H, CH), 3.98-4.12 (m, 4H, CH₂); $^{13}\text{C-NMR}$ (75 MHz) 12.2 (CH₃), 15.3 (CH₃), 16.2 (CH₃(Et)), 41.4 (d, $^1J_{\text{PC}} = 134$ Hz, CH), 46.6 (CH₃N), 61.9 (CH₂), 164.0 (C=N); $^{31}\text{P-NMR}$ (120 MHz) 28.7; IR 1637, 1453, 1392, 1050; MS (70 eV) 250 (M^+ , 34). Anal. Calcd. for C₁₀H₂₃N₂O₃P (250.15): C, 47.97; H, 9.27; N, 11.20. Found: C, 47.88; H, 9.24; N, 11.23.

General Procedure for the Preparation of the α,β -Unsaturated Hydrazones 1 from Functionalized Ylides 10. A dry flask, 100-mL, 2-necked, fitted with a dropping funnel, gas inlet, and magnetic stirrer, was charged 5 mmol of β -enehydrazino phosphonium salt 3 and 0.69 g (5 mmol) of potassium carbonate (K_2CO_3) and 40 mL of DMF. The mixture was allowed to stir for 1 h at room temperature. Then a solution 5 mmol of aldehyde in 10 mL of DMF was added at room temperature. The mixture was stirred until TLC indicated the disappearance of the aldehyde (1 day to 5 days). The mixture was diluted with 50 mL water and extracted with diethyl ether. The diethyl ether layer was washed with water. The organic layers were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by flash-chromatography.

***p*-Nitrophenylbuten-3-one *N,N*-dimethylhydrazone (1a).** 1049 mg (90 %) of 1a as a red oil ($R_f = 0.15$, hexane). Data for 1a: $^1\text{H-NMR}$ (250 MHz) 2.12 and 2.13 (s, 3H, *anti*- and *syn*-CH₃), 2.49 and 2.57 (s, 6H, *syn*- and *anti*-CH₃N), 6.93 (d, 1H, $^3J_{\text{HH}} = 16.4$ Hz, =CH), 7.49-8.16 (m, 5H, arom and HC=); $^{13}\text{C-NMR}$ (75 MHz) 13.6 and 20.2 (*anti*- and *syn*-CH₃), 47.1 and 48.1 (*anti*- and *syn*-CH₃N), 123.8-147.3 (C-arom, *syn*- and *anti*-C=C), 160.0 and 160.1 (*anti*- and *syn*-C=N); IR 2952, 2857, 1597,

1518, 1342; *MS* (70eV) 233 (M^+ , 100). Anal. Calcd. for $C_{12}H_{15}N_3O_2$ (233.12): C, 61.77; H, 6.48; N, 18.02. Found: C, 62.00; H, 6.47; N, 18.08.

2,4-Heptadien-6-one *N,N*-dimethylhydrazone (1b). 639 mg (84 %) of **1b** as a yellow oil ($R_f = 0.21$, hexane / diethyl ether, 10 / 1). Data for **1b**: 1H -NMR (250 MHz) 1.75 (t, 3H, $^3J_{HH} = 6.4$ Hz, CH_3), 1.99 (s, 3H, CH_3), 2.42 and 2.45 (s, 6H, *syn*- and *anti*- CH_3N), 5.72-6.72 (m, 4H, $HC=CH-CH=CH$); ^{13}C -NMR (75 MHz) 13.3 and 20.2 (*anti*- and *syn*- CH_3), 18.4 (CH_3), 47.3 and 47.9 (*anti*- and *syn*- CH_3N), 121.4-136.9 ($HC=CH-CH=CH$), 161.0 and 162.1 (*syn*- and *anti*- $C=N$); IR 2967, 1446, 1269, 1104, 1025; *MS* (70 eV) 152 (M^+ , 17). Anal. Calcd. for $C_9H_{16}N_2$ (152.13): C, 70.99; H, 10.60; N, 18.41. Found: C, 70.94; H, 10.56; N, 18.39.

Phenylbuten-3-one *N,N*-dimethylhydrazone (1c). 771 mg (82 %) of **1c** as a yellow oil ($R_f = 0.19$, hexane / diethyl ether, 10 / 1). Data for **1c**: 1H -NMR (250 MHz) 2.17 and 2.18 (s, 3H, *anti*- and *syn*- CH_3), 2.54 and 2.57 (s, 6H, *syn*- and *anti*- CH_3N), 6.88-7.47 (m, 6H, arom and $HC=$), 7.47 (d, 1H, $^3J_{HH} = 16.5$ Hz, $=CH$); ^{13}C -NMR (75 MHz) 13.3 and 20.2 (*anti*- and *syn*- CH_3), 47.1 and 47.8 (*anti*- and *syn*- CH_3N), 120.1-136.2 (C-arom, *syn*- and *anti*- $C=C$), 161.0 and 162.2 (*syn*- and *anti*- $C=N$); IR 2949, 2847, 1578, 1450; *MS* (70 eV) 188 (M^+ , 100). Anal. Calcd. for $C_{12}H_{16}N_2$ (188.13): C, 76.54; H, 8.57; N, 14.88. Found: C, 76.72; H, 8.60; N, 14.90.

***p*-Tolylbuten-3-one *N,N*-dimethylhydrazone (1d).** 819 mg (81 %) of **1d** as a yellow oil ($R_f = 0.17$, hexane). Data for **1d**: 1H -NMR (250 MHz) 2.17 and 2.18 (s, 3H, *anti*- and *syn*- CH_3), 2.33 (s, 3H, CH_3 -arom), 2.53 and 2.56 (s, 6H, *syn*- and *anti*- CH_3N), 6.86 (d, 1H, $^3J_{HH} = 16.5$ Hz, $=CH$), 6.89-7.46 (m, 5H, arom and $HC=$); ^{13}C -NMR (75 MHz) 13.4 and 20.4 (*anti*- and *syn*- CH_3), 21.3 (CH_3 -arom), 47.3 and 48.1 (*anti*- and *syn*- CH_3N), 119.4-139.1 (C-arom, *syn*- and *anti*- $C=C$), 161.4 and 162.7 (*syn*- and *anti*- $C=N$); IR 2950, 2855, 1618, 1580, 1518, 1460; *MS* (70 eV) 202 (M^+ , 74). Anal. Calcd. for $C_{13}H_{18}N_2$ (202.15): C, 77.17; H, 8.97; N, 13.85. Found: C, 76.91; H, 9.01; N, 13.89.

4-Methylfurfurylbuten-3-one *N,N*-dimethylhydrazone (1e). 778 mg (81 %) of **1e** as a yellow oil ($R_f = 0.18$, hexane / diethyl ether, 10 / 1). Data for **1e**: 1H -NMR (250 MHz) 2.08 and 2.09 (s, 3H, *anti*- and *syn*- CH_3), 2.25 and 2.30 (s, 3H, *anti*- and *syn*- CH_3 -Het), 2.51 and 2.52 (s, 6H, *syn*- and *anti*- CH_3N), 7.20 (d, 1H, $^3J_{HH} = 16.4$ Hz, $=CH$), 5.94-7.16 (m, 3H, $=CH-CH=$ and $HC=$); ^{13}C -NMR (75 MHz) 12.9 and 19.8 (*anti*- and *syn*- CH_3), 13.5 and 13.6 (*anti*- and *syn*- CH_3 -Het), 47.1 and 47.8 (*anti*- and *syn*- CH_3N), 107.9-153.6 ($C=CH-CH=C$ and *syn*- and *anti*- $C=C$), 160.8 and 161.9 (*syn*- and *anti*- $C=N$); IR 2947, 2848, 1629, 1585, 1450, 1365; *MS* (70 eV) 192 (M^+ , 78). Anal. Calcd. for $C_{11}H_{16}N_2O$ (192.13): C, 68.70; H, 8.39; N, 14.58. Found: C, 68.49; H, 8.36; N, 14.54.

Phenyl-3-hexen-5-one *N,N*-dimethylhydrazone (1f). 984 mg (91 %) of **1f** as a yellow oil ($R_f = 0.15$, hexane). Data for **1f**: 1H -NMR (250 MHz) 2.07 (s, 3H, CH_3), 2.47 and 2.54 (s, 6H, *syn*- and *anti*- CH_3N), 2.46-2.80 (m, 4H, CH_2), 6.83 (d, 1H, $^3J_{HH} = 16.1$ Hz, $=CH$), 6.20-7.30 (m, 6H, arom and $HC=$); ^{13}C -NMR (75 MHz) 13.4 and 20.4 (*anti*- and *syn*- CH_3), 34.7-35.3 (CH_2), 47.3 and 47.8 (*anti*- and *syn*- CH_3N), 123.7-142.0 (C-arom, *syn*- and *anti*- $C=C$), 161.4 and 163.0 (*syn*- and *anti*- $C=N$); IR 2945, 2850, 1645, 1607, 1588, 1500, 1456, 1367; *MS* (70 eV) 216 (M^+ , 43). Anal. Calcd. for $C_{14}H_{20}N_2$ (216.16): C, 77.72; H, 9.32; N, 12.96. Found: C, 77.97; H, 9.36; N, 12.91.

6-Methyl-3-hepten-2-one *N,N*-dimethylhydrazone (1g). 689 mg (82 %) of **1g** as a yellow oil ($R_f = 0.16$, hexane). Data for **1g**: 1H -NMR (250 MHz) 0.84-0.90 (m, 6H, CH_3), 1.59-1.70 (m, 1H, CH), 1.97-2.04 (m, 5H, CH_3 and CH_2), 2.45 (s, 6H, CH_3N), 6.06-6.16 (m, 1H, $HC=$), 6.70 (d, 1H, $^3J_{HH} = 16.1$ Hz, $=CH$); ^{13}C -NMR (75 MHz) 13.1 and 20.3 (*anti*- and *syn*- CH_3), 22.3 (CH_3), 28.2 (CH), 42.1 (CH_2), 47.1 and 47.6 (*anti*- and *syn*- CH_3N), 124.0 and 132.6 (*syn*- and *anti*- $HC=$), 135.9 and 138.7 (*anti*- and *syn*- $=CH$), 161.2 and 162.8 (*syn*- and *anti*- $C=N$); IR 2060, 2868, 1663, 1591, 1466, 1366; *MS* (70 eV) 168 (M^+ , 100). Anal. Calcd. for $C_{10}H_{20}N_2$ (168.16): C, 71.36; H, 11.98; N, 16.65. Found: C, 71.58; H, 11.97; N, 16.63.

***p*-Tolylbuten-3-one *N,N*-(*S*)-(-)-2-methoxymethylpirrolidinehydrazone (1h).** 1198 mg (88 %) of **1h** as a yellow oil ($R_f = 0.18$, hexane). Data for **1h**: 1H -NMR (250 MHz) 1.64-2.71 (m, 6H, ring CH_2), 2.10 and 2.11 (s, 3H, *anti*- and *syn*- CH_3), 2.32 (s, 3H, CH_3 -arom), 3.31-3.55 (m, 3H, CH_2O and CH), 3.36 (s, 3H, OCH_3), 6.88 (d, 1H, $^3J_{HH} = 16.5$ Hz, $=CH$), 6.73-7.42 (m, 5H, arom and $HC=$); ^{13}C -NMR (75 MHz) 14.8 and 20.5 (*anti*- and *syn*- CH_3), 21.3 (ring CH_2), 23.0 (CH_3 -arom), 27.1 (ring CH_2), 55.4 (ring CH_2), 59.2 (OCH_3), 66.8 (CH), 75.7 (CH_2O), 120.9-138.8 (C-arom, *syn*- and *anti*- $C=C$), 158.0 and 158.3 (*syn*- and *anti*- $C=N$); IR 2974, 2872, 1621, 1521, 1452, 1108; *MS* (70 eV) 272 (M^+ , 21). Anal. Calcd. for $C_{17}H_{24}N_2O$ (272.19): C, 74.95; H, 8.89; N, 10.29. Found: C, 75.22; H, 8.86; N, 10.26.

***p*-Tolylbuten-3-one *N,N*-(*R*)-(+)-2-methoxymethylpirrolidinehydrazone (1i).** 1020 mg (75 %) of **1i** as a yellow oil ($R_f = 0.20$, hexane / diethyl ether, 10 / 1). Data for **1i**: 1H -NMR (250 MHz) 1.80-2.54 (m, 6H, ring CH_2), 2.10 (s, 3H, CH_3), 2.32 (s, 3H, CH_3 -arom), 3.28-3.53 (m, 3H, CH_2O and CH), 3.36 (s, 3H, OCH_3), 6.88 (d, 1H, $^3J_{HH} = 16.5$ Hz, $=CH$), 6.81-7.36 (m, 5H, arom and $HC=$); ^{13}C -NMR (75 MHz) 14.8 (CH_3), 21.3 (ring CH_2), 23.0 (CH_3 -arom), 27.1 (ring CH_2), 55.3

(ring CH₂), 59.2 (OCH₃), 66.8 (CH), 75.7 (CH₂O), 126.7-137.9 (C-arom and C=C), 158.3 (C=N); IR 2973, 2872, 1618, 1519, 1452, 1107; MS (70 eV) 272 (M⁺, 3). Anal. Calcd. for C₁₇H₂₄N₂O (272.19): C, 74.95; H, 8.89; N, 10.29. Found: C, 75.13; H, 8.87; N, 10.27.

Ethyl-4-dimethylhydrazone-E-2-pentenoate (1j). 736 mg (80 %) of **1j** as a yellow oil (*R_f* = 0.21, hexane / diethyl ether, 10 / 1). Data for **1j**: ¹H-NMR (250 MHz) 1.24 (t, 3H, ³J_{HH} = 7.1 Hz, CH₃(Et)), 2.00 (s, 3H, CH₃), 2.61 (s, 6H, CH₃N), 4.16 (q, 2H, ³J_{HH} = 7.1 Hz, OCH₂), 6.06 (d, 1H, ³J_{HH} = 16 Hz, HC=), 7.24 (d, 1H, ³J_{HH} = 16 Hz, =CH); ¹³C-NMR (75 MHz) 13.5 (CH₃(Et)), 13.6 (CH₃), 46.6 (CH₃N), 59.8 (OCH₂), 120.9 (HC=), 145.2 (=CH), 155.8 (C=N), 165.6 (C=O); IR 2960, 2868, 1723, 1637, 1473, 1262; MS (70 eV) 184 (M⁺, 23). Anal. Calcd. for C₉H₁₆N₂O₂ (184.12): C, 58.66; H, 8.76; N, 15.21. Found: C, 58.78; H, 8.73; N, 15.16.

General Procedure for the Preparation of the α,β -Unsaturated Hydrazones 1 from Functionalized Phosphine Oxides 8 (6), or from phosphonates 9 (7). A dry flask, 100-mL, 2-necked, fitted with a dropping funnel, gas inlet, and magnetic stirrer, was charged 5 mmol of methyl lithium or 5 mmol of lithium diisopropylamide (LDA) and 40 mL of THF. The temperature was allowed to descend to -78 °C and a solution 5 mmol of compounds 6, 7, 8 or 9 in 40 mL of THF was then added. The mixture was allowed to stir at this temperature for 1 h. A solution 5 mmol of carbonyl compound (aldehydes or ketones) in 10 mL of THF was added at -78 °C. The mixture was stirred until TLC indicated the disappearance of the carbonyl compound (14 h to 3 days). The mixture was diluted with 50 mL water and extracted with CH₂Cl₂. The CH₂Cl₂ layers were washed with water. The organic layers were dried over MgSO₄, filtered, and concentrated. The crude product was purified by flash-chromatography.

p-Tolylpenten-3-one N,N-dimethylhydrazone (1k). 919 mg (85 %) of **1k** as a yellow oil (*R_f* = 0.17, hexane). Data for **1k**: ¹H-NMR (250 MHz) 1.14-1.21 (m, 3H, CH₃), 2.32 (s, 3H, CH₃-arom), 2.52 (s, 6H, CH₃N), 2.55-2.70 (m, 2H, CH₂), 6.72 (d, 1H, ³J_{HH} = 16.6 Hz, =CH), 6.89-7.42 (m, 5H, arom and HC=); ¹³C-NMR (75 MHz) 12.4 (CH₃), 20.1 (CH₃-arom), 21.0 (CH₂), 47.8 (CH₃N), 118.4-141.6 (C-arom and C=C), 168.9 (C=N); IR 2987, 2855, 1609, 1511, 1465; MS (70 eV) 216 (M⁺, 100). Anal. Calcd. for C₁₄H₂₀N₂ (216.16): C, 77.72; H, 9.32; N, 12.96. Found: C, 77.67; H, 9.33; N, 13.00.

Diphenylbuten-3-one N,N-dimethylhydrazone (1l). 1057 mg (80 %) of **1l** as a yellow oil (*R_f* = 0.20, hexane / diethyl ether, 10 / 1). Data for **1l**: ¹H-NMR (250 MHz) 1.57 (s, 3H, anti- and syn-CH₃), 2.54 and 2.66 (s, 6H, anti- and syn-CH₃N), 6.78 and 6.95 (s, 1H, anti- and syn-HC=), 7.19-7.37 (m, 10H, arom); ¹³C-NMR (75 MHz) 18.1 and 22.4 (anti- and syn-CH₃), 47.0 and 47.9 (syn- and anti-CH₃N), 123.3-146.2 (C-arom, syn- and anti-HC=C), 160.5 and 162.4 (syn- and anti-C=N); IR 2947, 2843, 1603, 1493, 1445, 1350; MS (70 eV) 264 (M⁺, 100). Anal. Calcd. for C₁₈H₂₀N₂ (264.16): C, 81.77; H, 7.63; N, 10.60. Found: C, 81.80; H, 7.62; N, 10.58.

Cyclohexylidenpropen-2-one N,N-dimethylhydrazone (1m). 775 mg (86 %) of **1m** as a yellow oil (*R_f* = 0.15, hexane). Data for **1m**: ¹H-NMR (250 MHz) 1.89 and 1.92 (s, 3H, anti- and syn-CH₃), 2.34 (s, 6H, CH₃N), 1.41-2.30 (m, 10H, ring CH₂), 5.53 and 5.83 (s, 1H, anti- and syn-HC=); ¹³C-NMR (75 MHz) 18.2 (CH₃), 24.1-37.7 (ring CH₂), 46.8 (CH₃N), 118.2 and 121.8 (syn- and anti-HC=), 145.7 and 147.2 (anti- and syn=C), 160.3 and 162.4 (syn- and anti-C=N); IR 2955, 2857, 1618, 1512, 1446; MS (70 eV) 180 (M⁺, 22). Anal. Calcd. for C₁₁H₂₀N₂ (180.16): C, 73.27; H, 11.19; N, 15.54. Found: C, 73.32; H, 11.15; N, 15.52.

4,6-Dimethyl-3-hepten-2-one N,N-dimethylhydrazone (1n). 619 mg (68 %) of **1n** as a yellow oil (*R_f* = 0.18, hexane / diethyl ether, 10 / 1). Data for **1n**: ¹H-NMR (250 MHz) 0.77-0.99 (m, 6H, CH₃(i-Bu)), 1.75-2.03 (m, 3H, CH₂ and CH), 2.05 and 2.09 (s, 3H, anti- and syn-CH₃), 2.48 and 2.50 (s, 6H, syn- and anti-CH₃N), 5.72 and 6.09 (s, 1H, anti- and syn-HC=); ¹³C-NMR (75 MHz) 18.5 (CH₃), 22.5 (CH₃(i-Bu)), 26.1 (CH), 41.8 (CH₂), 47.2 (CH₃N), 122.4 (HC=), 126.0 (C=), 162.8 (C=N); IR 2960, 1650, 1473, 1387; MS (70 eV) 182 (M⁺, 19). Anal. Calcd. for C₁₁H₂₂N₂ (182.18): C, 72.46; H, 12.17; N, 15.37. Found: C, 72.56; H, 12.13; N, 15.41.

Phenylbuten-3-one N,N-(S)-(-)-2-methoxymethylpiperolidinehydrazone (1o). 1201 mg (93 %) of **1o** as a yellow oil (*R_f* = 0.19, hexane). Data for **1o**: ¹H-NMR (250 MHz) 1.84-2.70 (m, 6H, ring CH₂), 2.11 (s, 3H, CH₃), 3.32-3.53 (m, 3H, CH₂O and CH), 3.36 (s, 3H, OCH₃), 6.85 (d, 1H, ³J_{HH} = 16.5 Hz, =CH), 7.24-7.48 (m, 6H, arom and HC=); ¹³C-NMR (75 MHz) 14.9 (CH₃), 23.1 (ring CH₂), 27.1 (ring CH₂), 55.4 (ring CH₂), 59.2 (OCH₃), 66.8 (CH), 75.7 (CH₂O), 126.8-136.8 (C-arom and C=C), 157.8 (C=N); IR 2973, 2874, 1722, 1669, 1449, 1110; MS (70 eV) 258 (M⁺, 18). Anal. Calcd. for C₁₆H₂₂N₂O (258.17): C, 74.37; H, 3.59; N, 10.85. Found: C, 74.12; H, 3.60; N, 10.88.

p-Anisylbuten-3-one N,N-dimethylhydrazone (1p). 807 mg (74 %) of **1p** as a yellow oil (*R_f* = 0.20, hexane / diethyl ether, 10 / 1). Data for **1p**: ¹H-NMR (250 MHz) 2.00 and 2.02 (s, 3H, anti- and syn-CH₃), 2.40 and 2.41 (s, 6H, syn- and anti-CH₃N), 3.59 and 3.60 (s, 3H, anti- and syn-OCH₃), 6.65 (d, 1H, ³J_{HH} = 16.5 Hz, =CH), 6.67-7.31 (m, 5H, arom and HC=);

$^{13}\text{C-NMR}$ (75 MHz) 13.3 and 20.3 (*anti*- and *syn*- CH_3), 47.3 and 48.0 (*anti*- and *syn*- CH_3N), 55.2 and 55.3 (*anti*- and *syn*- OCH_3), 114.1-143.1 (C-*arom*, *syn*- and *anti*- $\text{C}=\text{C}$), 159.8 and 162.8 (*anti*- and *syn*- $\text{C}=\text{N}$); *IR* 2955, 2861, 1604, 1511, 1249, 1175; *MS* (70 eV) 218 (M^+ , 92). Anal. Calcd. for $\text{C}_{13}\text{H}_{18}\text{N}_2\text{O}$ (218.14): C, 71.51; H, 8.32; N, 12.84. Found: C, 71.67; H, 8.29; N, 12.80.

2-Methyl-*p*-tolylbuten-3-one *N,N*-dimethylhydrazone (1q). 994 mg (92 %) from **8** and 713 mg (66 %) from **6** of **1q** as a yellow oil ($R_f = 0.18$, hexane / diethyl ether, 10 / 1). Data for **1q**: $^1\text{H-NMR}$ (250 MHz) 2.13 (s, 3H, CH_3), 2.22 (s, 3H, CH_3), 2.37 (s, 3H, CH_3 -*arom*), 2.57 (s, 6H, CH_3N), 6.88 (s, 1H, $=\text{CH}$), 7.17 (d, 2H, $^4J_{\text{HH}} = 8$ Hz, *arom*), 7.25 (d, 2H, $^4J_{\text{HH}} = 8$ Hz, *arom*); $^{13}\text{C-NMR}$ (75 MHz) 14.3 (CH_3), 14.7 (CH_3), 21.3 (CH_3 -*arom*), 47.4 (CH_3N), 127.1-138.8 (C-*arom* and $\text{C}=\text{C}$), 164.2 ($\text{C}=\text{N}$); *IR* 2955, 2858, 1510, 1447, 1365; *MS* (70 eV) 216 (M^+ , 60). Anal. Calcd. for $\text{C}_{14}\text{H}_{20}\text{N}_2$ (216.16): C, 77.72; H, 9.33; N, 12.96. Found: C, 77.79; H, 9.26; N, 12.95.

2-Allyl-*p*-tolylbuten-3-one *N,N*-dimethylhydrazone (1r). 872 mg (72 %) from **8** and 690 mg (57 %) from **6** of **1r** as a yellow oil ($R_f = 0.16$, hexane). Data for **1r**: $^1\text{H-NMR}$ (250 MHz) 2.21 (s, 3H, CH_3), 2.36 (s, 3H, CH_3 -*arom*), 2.55 (s, 6H, CH_3N), 3.38-3.41 (m, 2H, CH_2), 5.00-5.08 (m, 2H, $\text{CH}_2=$), 6.00-6.06 (m, 1H, $=\text{CH}$), 6.94 (s, 1H, $=\text{CH}$), 7.15-7.34 (m, 4H, *arom*); $^{13}\text{C-NMR}$ (75 MHz) 14.6 (CH_3), 21.2 (CH_3 -*arom*), 31.9 (CH_2), 47.4 (CH_3N), 114.9 ($\text{CH}_2=$), 128.9-139.3 (C-*arom* and $\text{C}=\text{CH}$), 137.4 ($=\text{CH}$), 162.4 ($\text{C}=\text{N}$); *IR* 2955, 2856, 1670, 1511, 1567; *MS* (70 eV) 242 (M^+ , 28). Anal. Calcd. for $\text{C}_{16}\text{H}_{22}\text{N}_2$ (242.18): C, 79.28; H, 9.15; N, 11.56. Found: C, 79.28; H, 9.18; N, 11.54.

2-Methyl-*p*-anisylbuten-3-one *N,N*-dimethylhydrazone (1s). 882 mg (76 %) from **9** and 696 mg (60 %) from **7** of **1s** as a yellow oil ($R_f = 0.20$, hexane / diethyl ether, 10 / 1). Data for **1s**: $^1\text{H-NMR}$ (250 MHz) 2.11 (s, 3H, CH_3), 2.21 (s, 3H, CH_3), 2.55 (s, 6H, CH_3N), 3.82 (s, 3H, OCH_3), 6.88 (s, 1H, $=\text{CH}$), 6.92-7.32 (m, 4H, *arom*); $^{13}\text{C-NMR}$ (75 MHz) 14.2 (CH_3), 14.6 (CH_3), 47.4 (CH_3N), 55.3 (OCH_3), 113.6-130.7 (C-*arom* and $\text{C}=\text{CH}$), 164.1 ($\text{C}=\text{N}$); *IR* 2955, 2853, 1606, 1519, 1251, 1178; *MS* (70 eV) 232 (M^+ , 94). Anal. Calcd. for $\text{C}_{14}\text{H}_{20}\text{N}_2\text{O}$ (232.16): C, 72.36; H, 8.68; N, 12.06. Found: C, 72.59; H, 8.66; N, 12.02.

Reaction of β -*N*-phenylhydrazonopropylidiphenylphosphine oxide **6e with *p*-tolyl aldehyde.** A dry flask, 100-mL, 2-necked, fitted with a dropping funnel, gas inlet, and magnetic stirrer, was charged 10 mmol of lithium diisopropylamide (*LDA*) and 50 mL of *THF*. The temperature was allowed to descend to -78°C and a solution 1.74 g (5 mmol) of phosphine oxide **6e** in 40 mL of *THF* was then added. The mixture was allowed to stir for 1 h. A solution 0.6 g (5 mmol) of *p*-tolyl aldehyde in 10 mL of *THF* was added at -78°C . The mixture was stirred at room temperature until *TLC* indicated the disappearance of the carbonyl compound (17 h). The mixture was diluted with 50 mL water and extracted with CH_2Cl_2 . The CH_2Cl_2 layers were washed with water. The organic layers were dried over MgSO_4 , filtered, and concentrated.

***p*-Tolylbuten-3-one *N*-phenylhydrazone (13).** The crude product was purified by flash-chromatography (hexane / diethyl ether, 10 / 1) to afford 875 mg (70 %) of **13** as a yellow solid. Data for **13**: mp 165-167 $^\circ\text{C}$; $^1\text{H-NMR}$ (250 MHz) 2.08 (s, 3H, CH_3), 2.36 (s, 3H, CH_3 -*arom*), 6.73 (d, 1H, $^3J_{\text{HH}} = 16.4$ Hz, $=\text{CH}$), 6.87-7.41 (m, 10H, *arom*, $\text{HC}=\text{}$ and NH); $^{13}\text{C-NMR}$ (75 MHz) 10.1 (CH_3), 21.3 (CH_3 -*arom*), 113.2-144.7 (C-*arom* and $\text{C}=\text{C}$), 144.7 ($\text{C}=\text{N}$); *IR* (*KBr*) 3348, 2927, 2855, 1618, 1506, 1249, 1137; *MS* (70 eV) 250 (M^+ , 100). Anal. Calcd. for $\text{C}_{17}\text{H}_{18}\text{N}_2$ (250.15): C, 81.55; H, 7.25; N, 11.20. Found: C, 81.57; H, 7.27; N, 11.20.

Synthesis of 1-Phenyl-3-methyl-5-*p*-tolylpyrazole (15).

1.25 g (5 mmol) of *p*-tolylbuten-3-one *N*-phenylhydrazone **13** were heated at 100°C in 15 mL of Toluene until *TLC* indicated the disappearance of the compound **13**. The mixture was concentrated and the crude product was purified by flash-chromatography (hexane / diethyl ether, 10 / 1) to afford 1191 mg (96 %) of **15** as a yellow oil ($R_f = 0.19$). Data for **15**: $^1\text{H-NMR}$ (250 MHz) 2.34 (s, 3H, CH_3), 2.38 (s, 3H, CH_3 -*arom*), 6.28 (s, 1H, $\text{HC}=\text{}$), 6.91-7.34 (m, 9H, *arom*); $^{13}\text{C-NMR}$ (75 MHz) 13.6 (CH_3), 21.3 (CH_3 -*arom*), 107.5 ($\text{HC}=\text{}$), 124.5-143.9 (C-*arom* and $=\text{C}$), 149.4 ($\text{C}=\text{N}$); *IR* 3026, 2921, 1598, 1499, 1368; *MS* (70 eV) 248 (M^+ , 100). Anal. Calcd. for $\text{C}_{17}\text{H}_{16}\text{N}_2$ (248.13): C, 82.21; H, 6.50; N, 11.29. Found: C, 82.23; H, 6.52; N, 11.25.

Reaction of β -*N*-phenylhydrazonopropylidiphenylphosphine oxide **6e with diphenyl ketone.** This reaction was performed like synthesis of compound **13** with diphenyl ketone 0.91 g (5 mmol) and was refluxed for 3 days. **1,5,5-triphenyl-3-methyl-4,5-dihydropyrazole (14)**. The crude product was purified by flash-chromatography (hexane) to afford 999 mg (64 %) of **14** as a white solid. An analytical sample was obtained by recrystallization from CHCl_3 / hexane. Data for **14**: mp 193-194 $^\circ\text{C}$; $^1\text{H-NMR}$ (250 MHz) 2.10 (s, 3H, CH_3), 3.66 (s, 2H, CH_2), 6.59-7.54 (m, 15H, *arom*); $^{13}\text{C-NMR}$ (75 MHz) 15.8 (CH_3), 61.4 (CH_2), 114.9-128.3 (C-*arom* and $\text{C}(\text{Ph})_2$), 146.9 ($\text{C}=\text{N}$); *IR* (*KBr*) 2920, 2835, 1597, 1498, 1353; *MS* (70 eV) 312 (M^+ , 100). Anal. Calcd. for $\text{C}_{22}\text{H}_{20}\text{N}_2$ (312.16): C, 84.57; H, 6.46; N, 8.97. Found: C, 84.61; H, 6.44; N, 8.94.

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